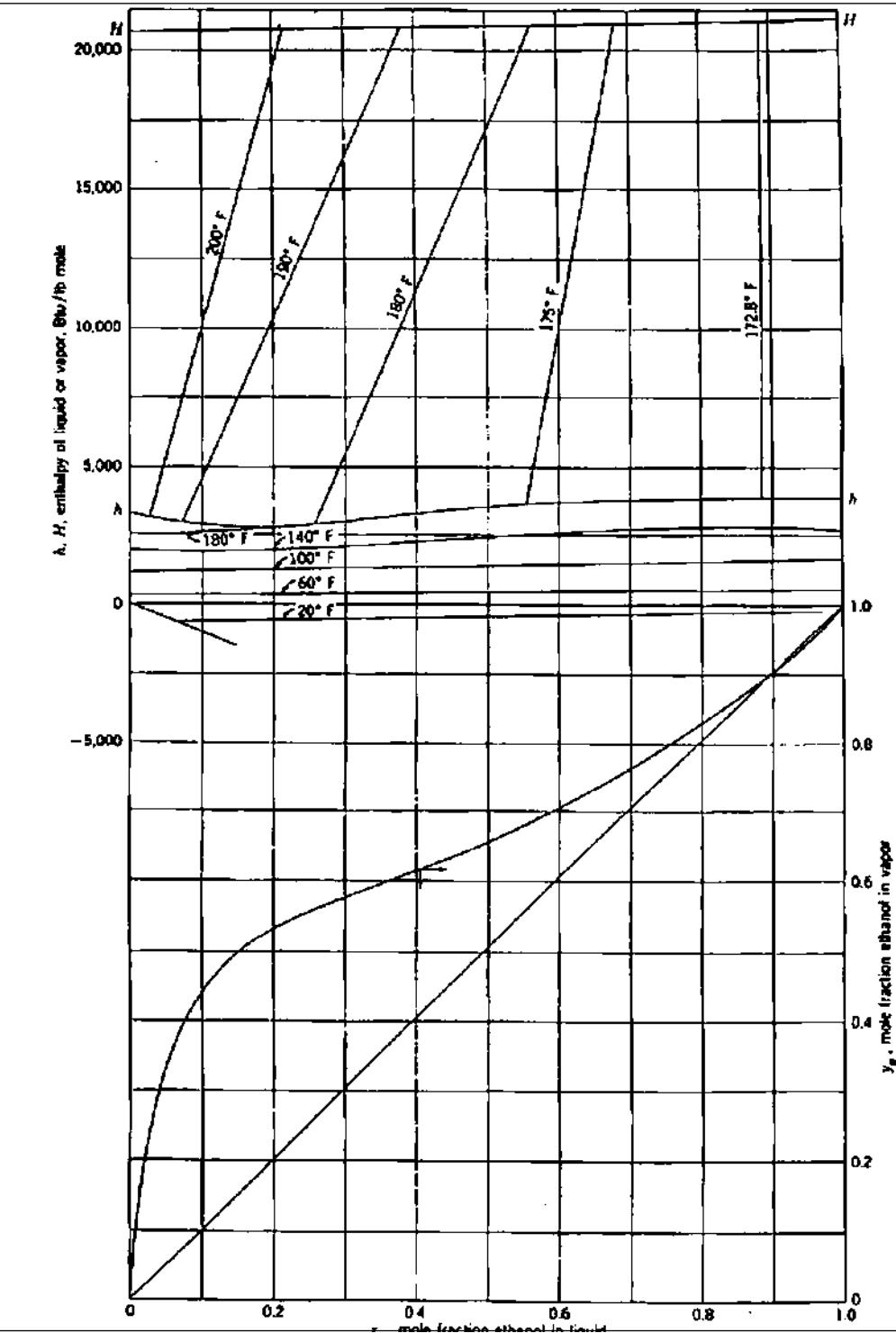


Acqua-Alcool Etilico

I atm

$$^{\circ}\text{C} = (^{\circ}\text{F}-32) \cdot 0.556$$

| Btu/lb=2.3 KJ/Kg



Acqua-Alcool Etilico
1 atm

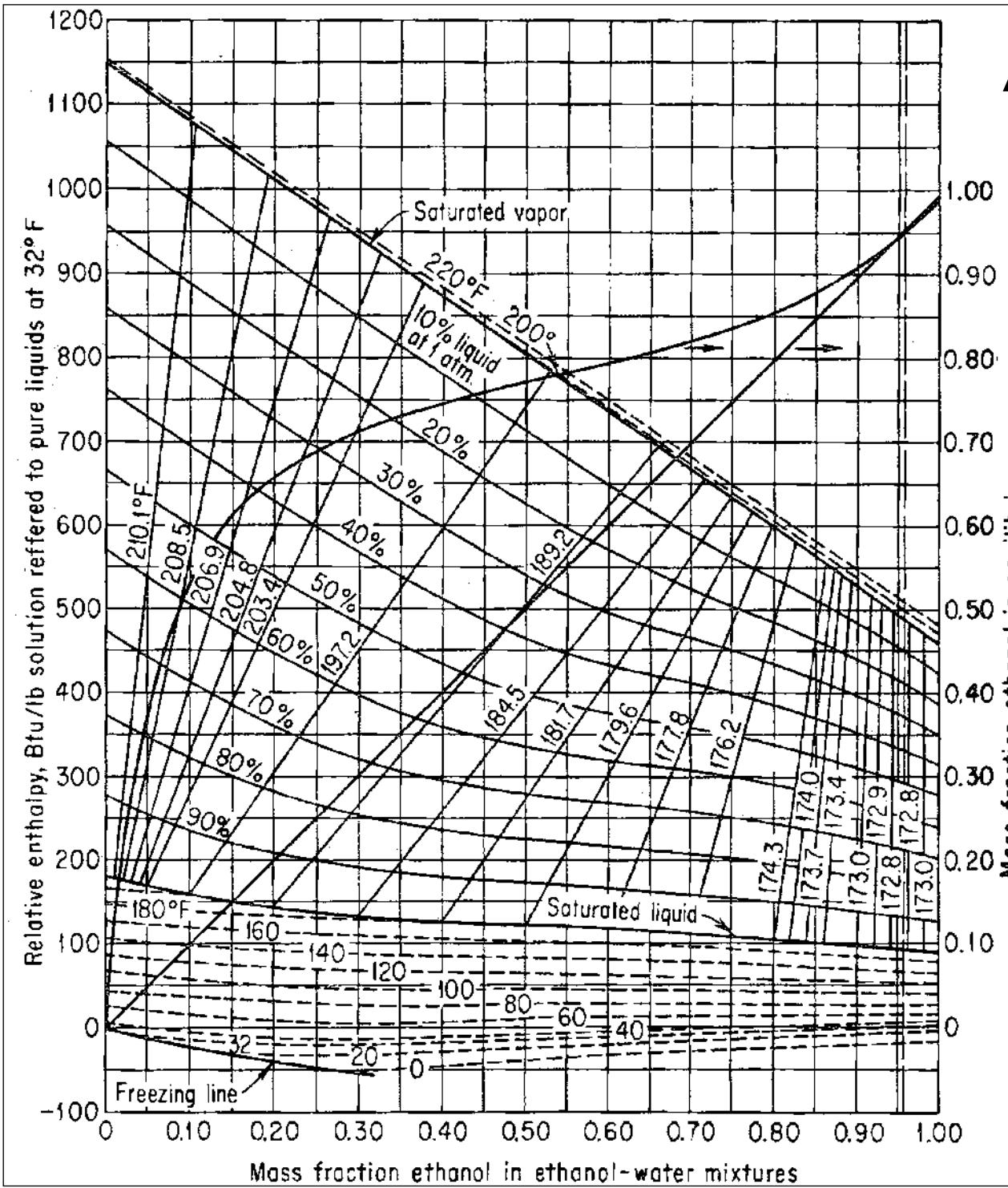
$$^{\circ}\text{C} = (^{\circ}\text{F}-32) 0.556$$

$$1 \text{ Btu/lb} = 2.3 \text{ KJ/Kg}$$

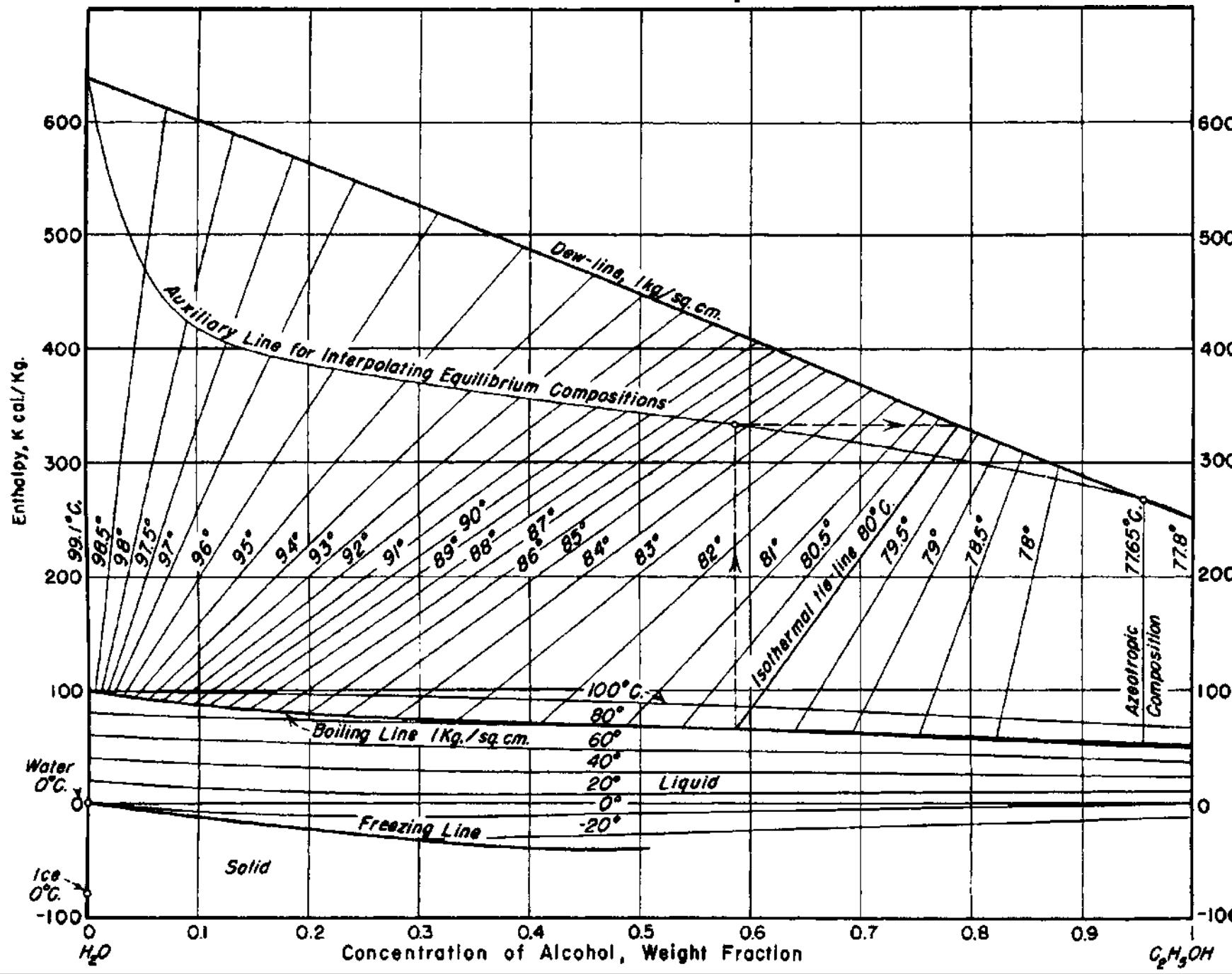
Acqua-Alcool Etilico 1 atm

$$^{\circ}\text{C} = (^{\circ}\text{F}-32) 0.556$$

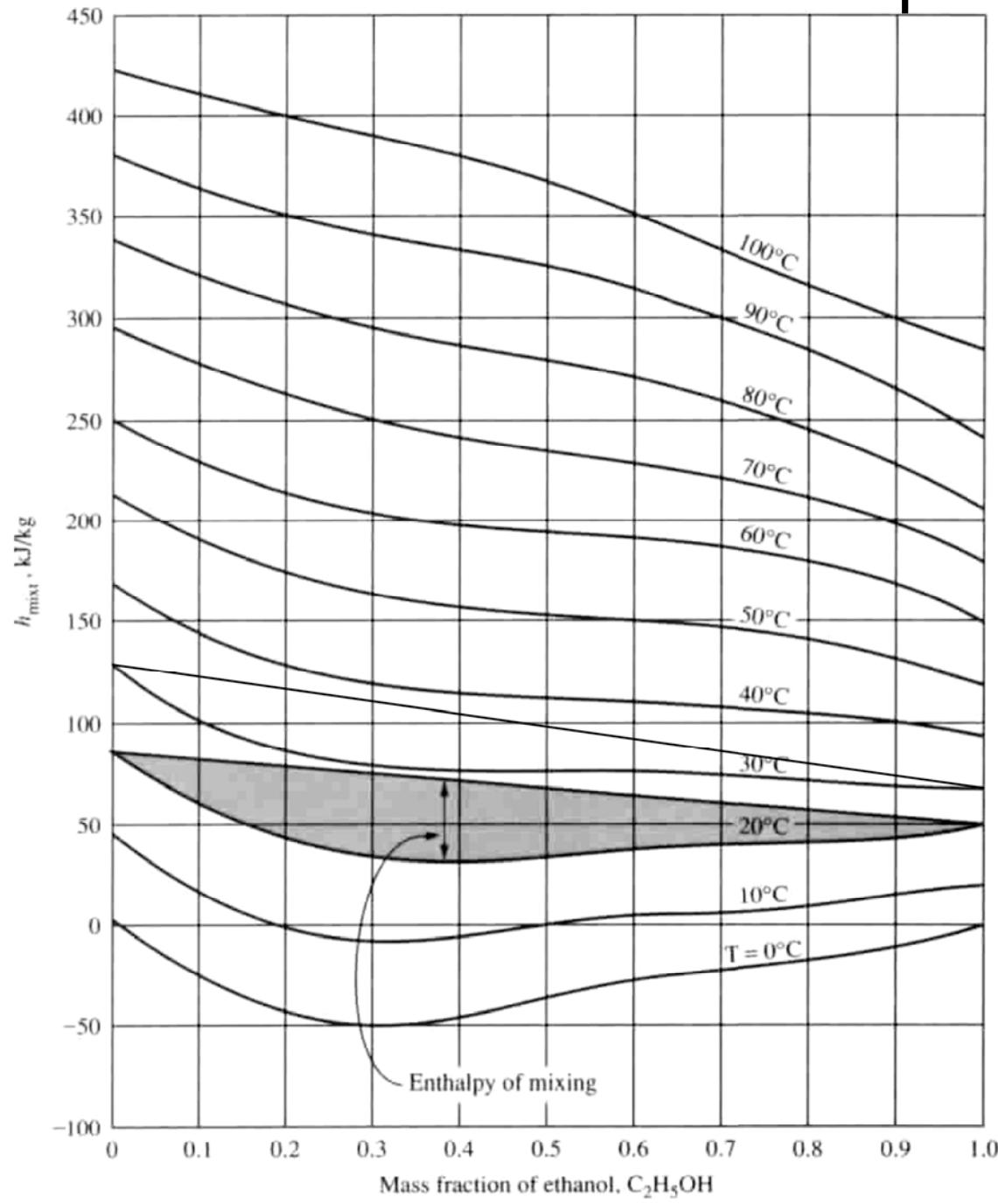
$$1 \text{ Btu/lb} = 2.3 \text{ KJ/Kg}$$



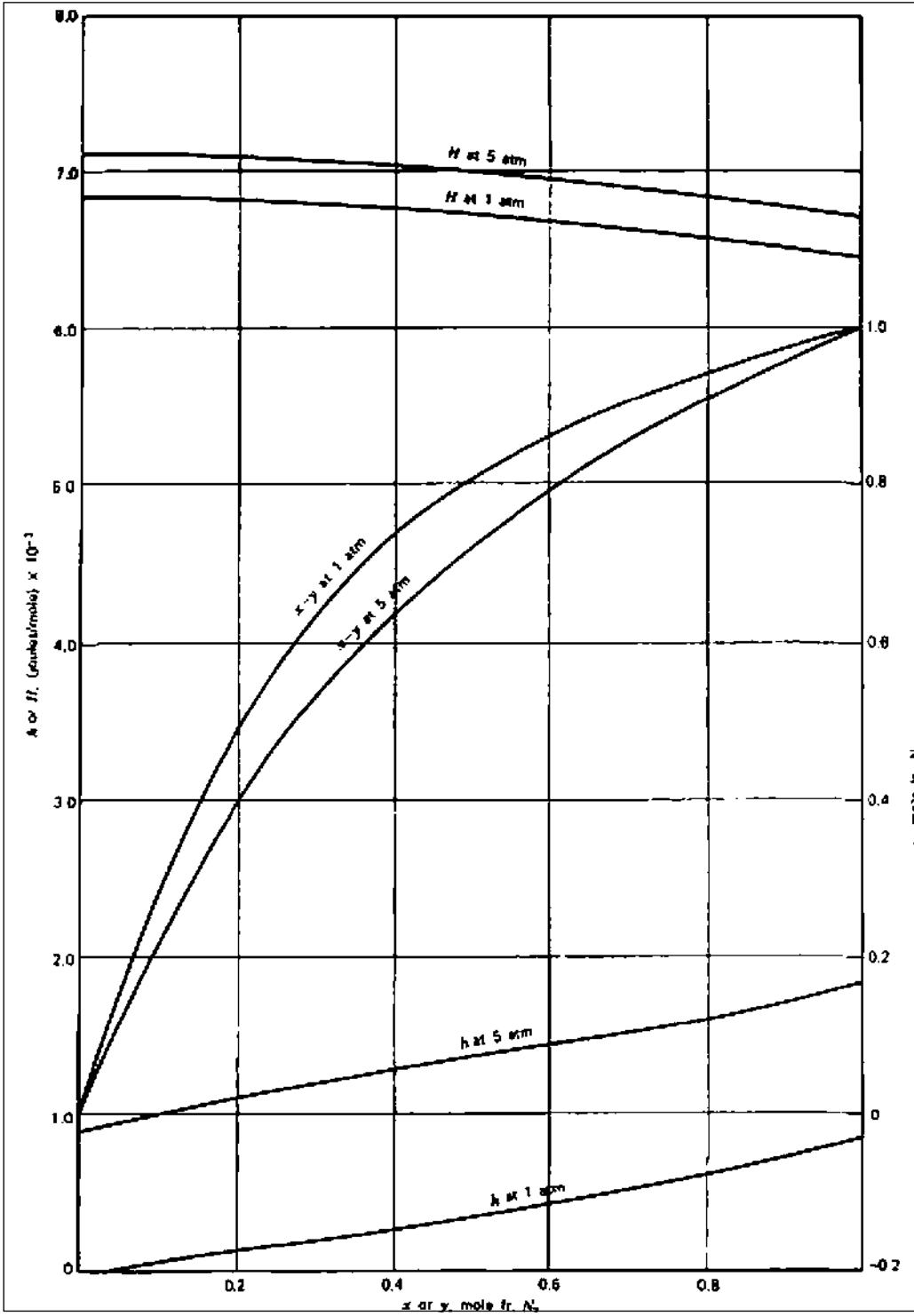
Acqua-Alcool Etilico ad 1 atm



Acqua-Alcool Etilico ad 1 atm

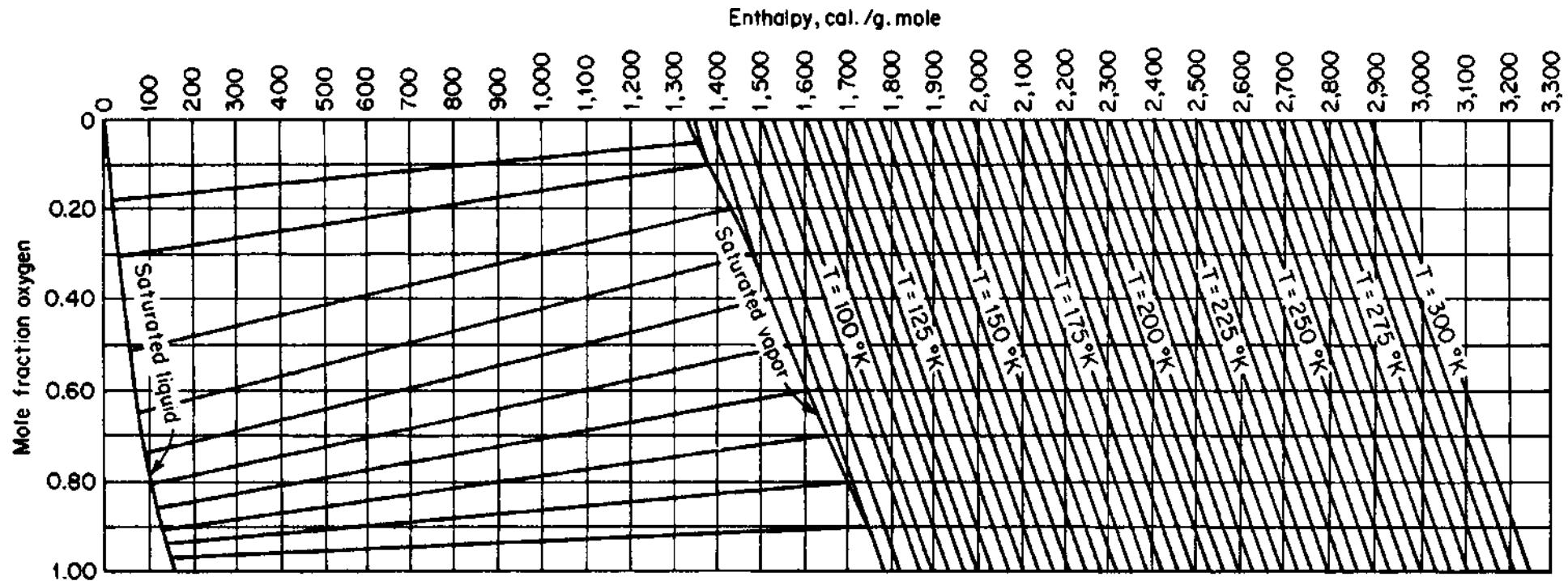


Ossigeno-Azoto 1 atm e 5 atm



Ossigeno-Azoto

| atm

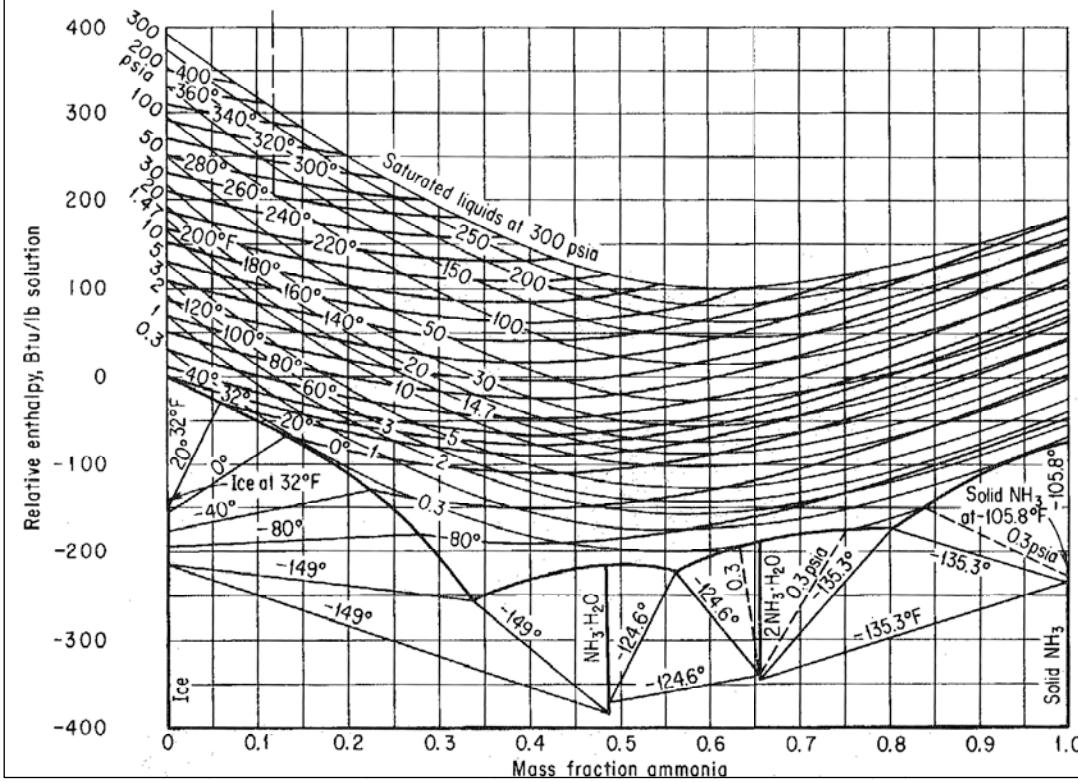
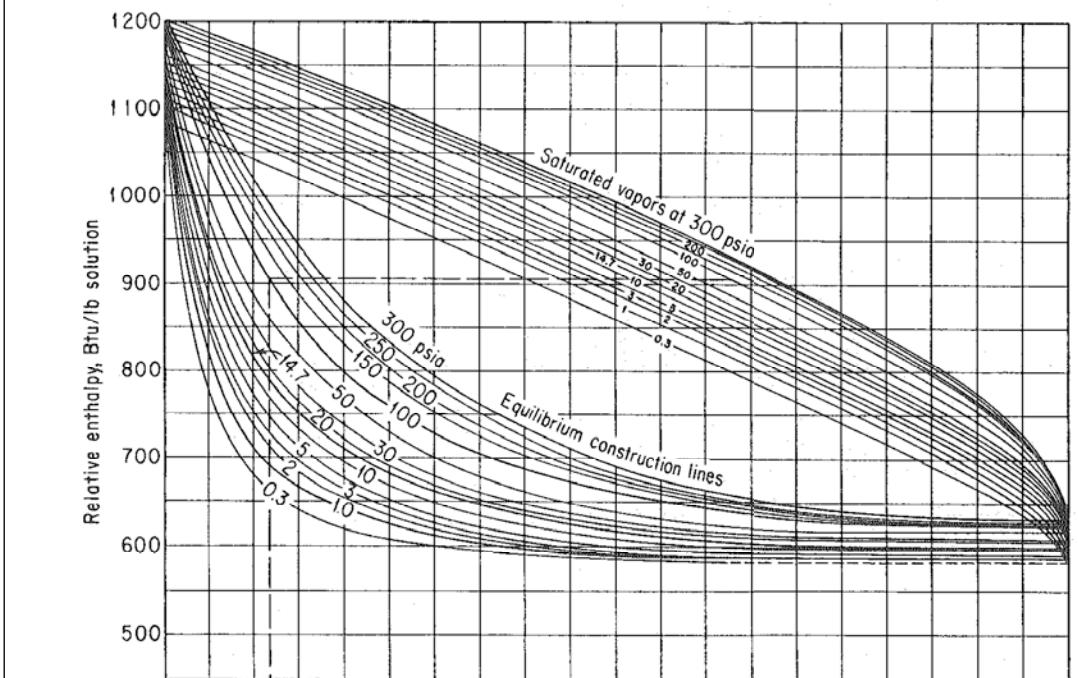


Acqua-Ammoniaca

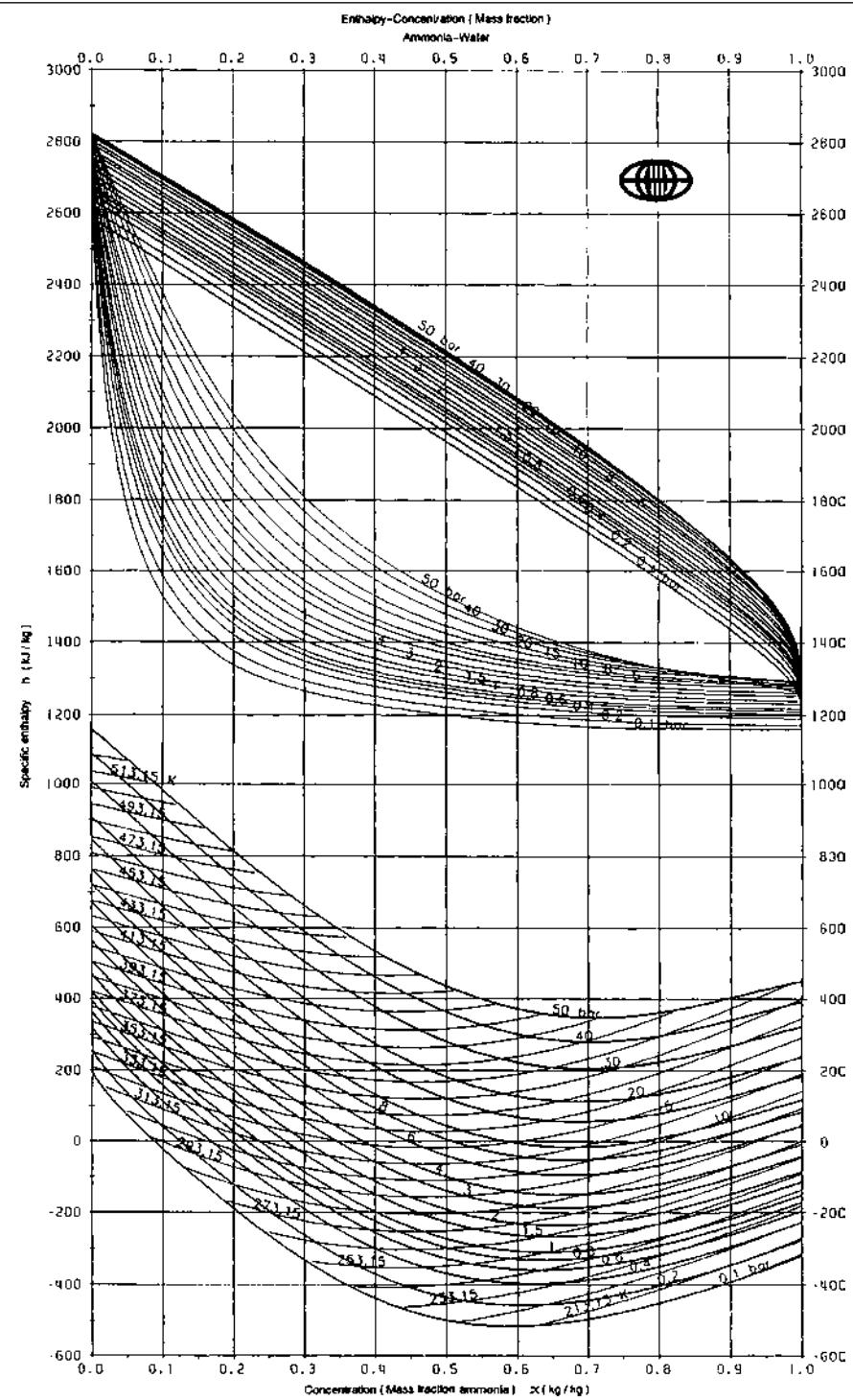
1 psia=0.0689 bar

°C=(°F-32) 0.556

1 Btu/lb=2.3 KJ/Kg

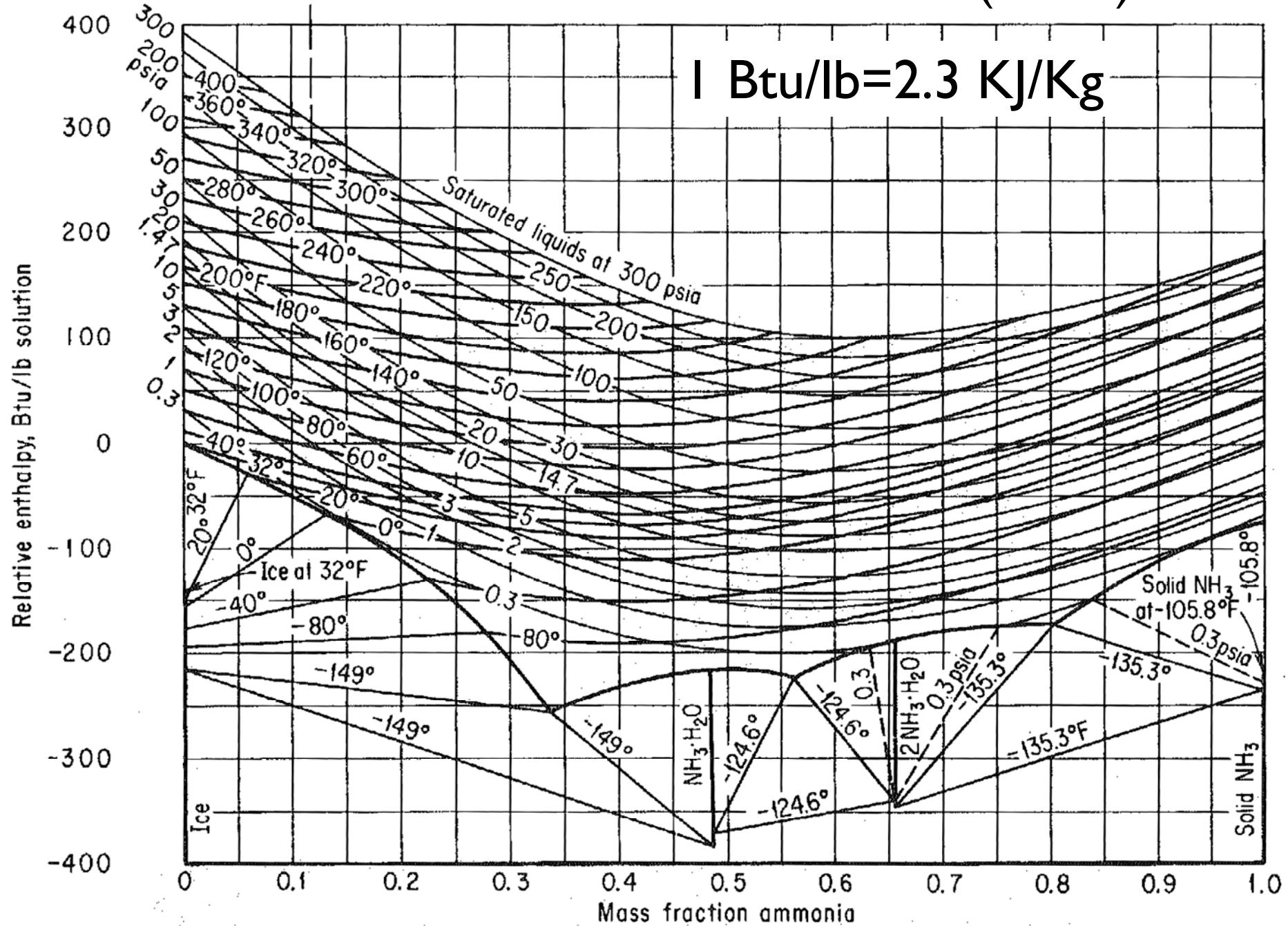


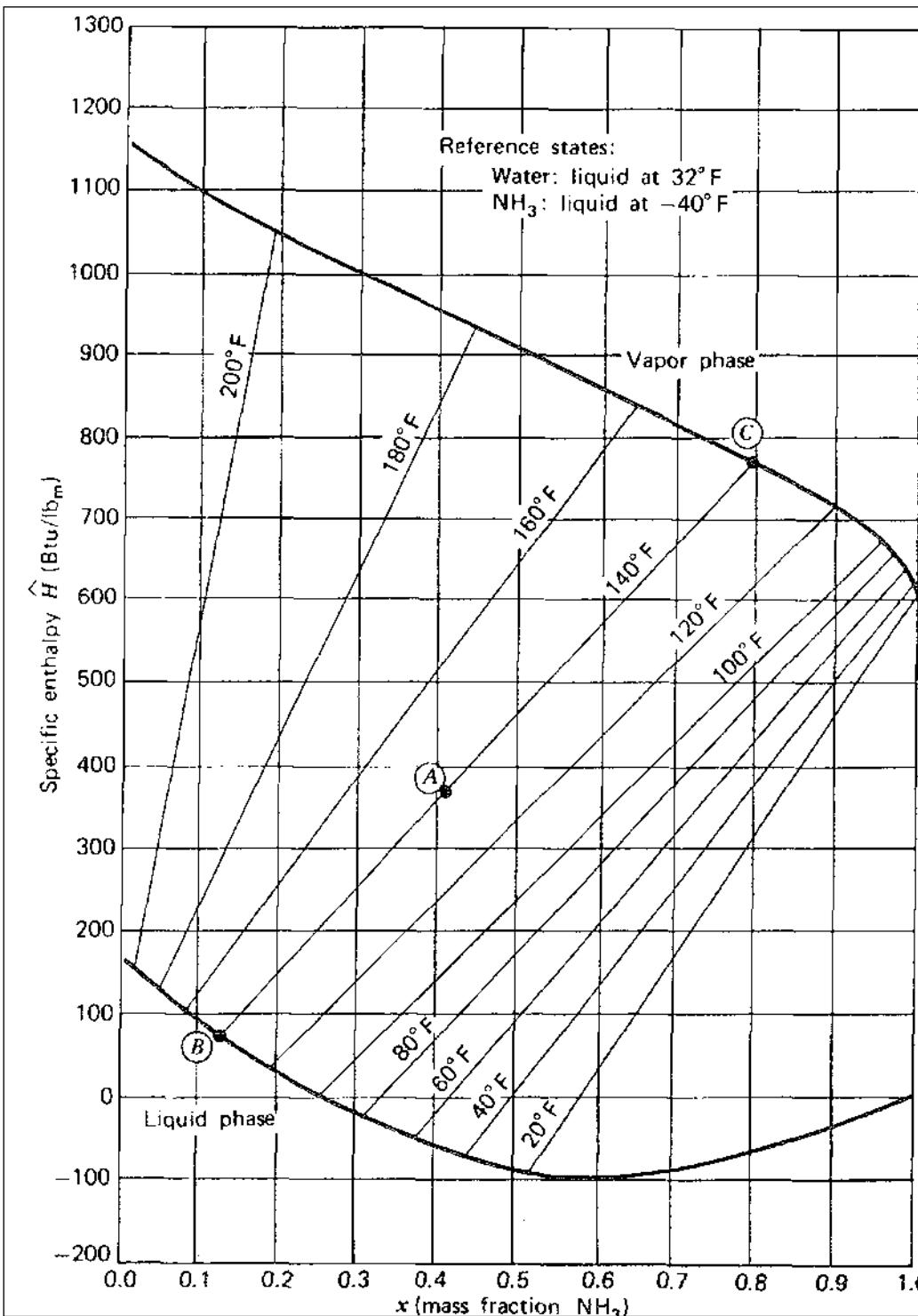
Acqua-Ammoniaca



Acqua-Ammoniaca | psia=0.0689 bar | °C=(°F-32) 0.556

| Btu/lb=2.3 KJ/Kg





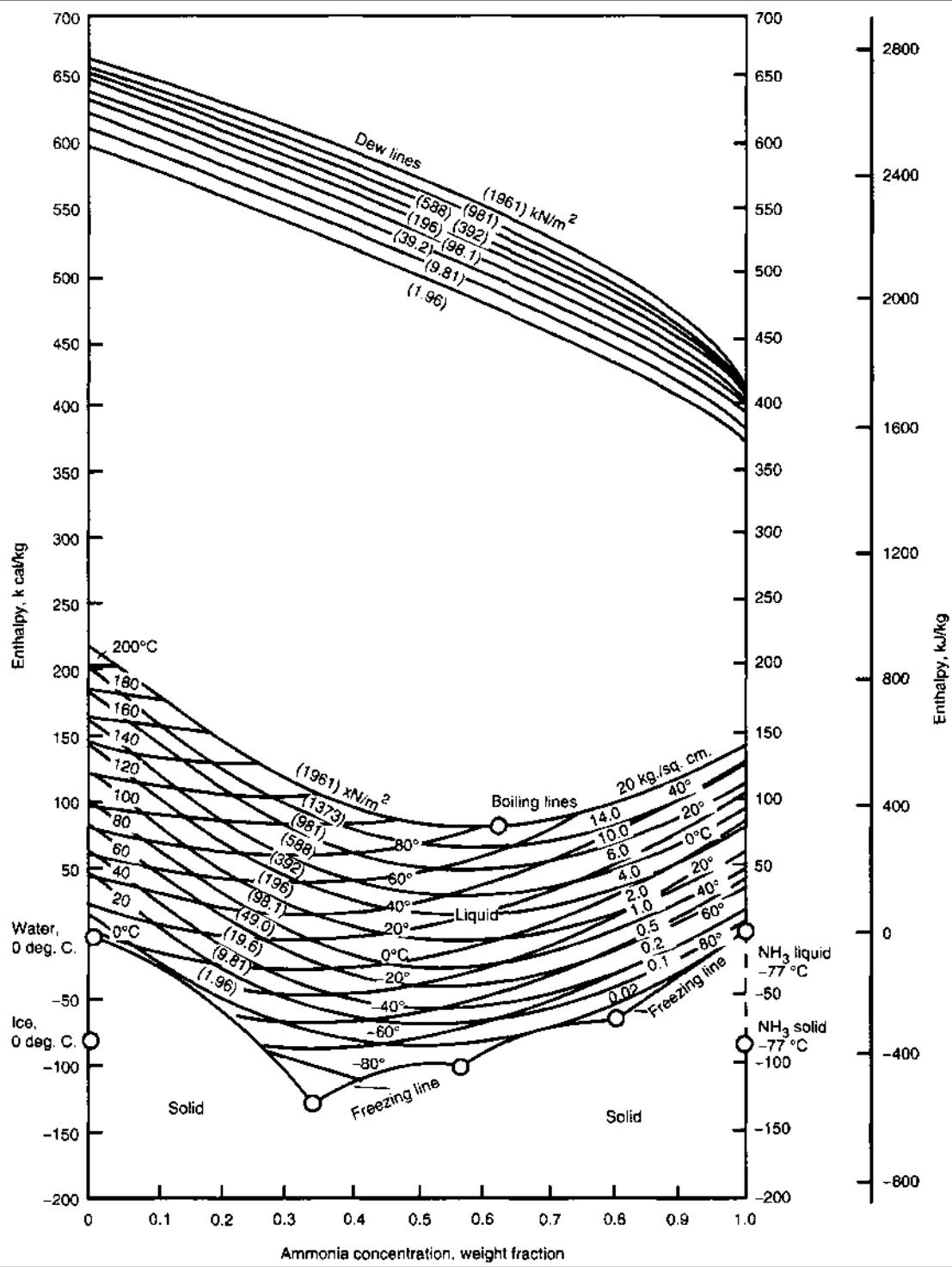
Acqua-Ammoniaca

$$\text{Ipsia} = 0.0689 \text{ bar}$$

$$^{\circ}\text{C} = (^{\circ}\text{F} - 32) \cdot 0.556$$

$$\text{I Btu/lb} = 2.3 \text{ KJ/Kg}$$

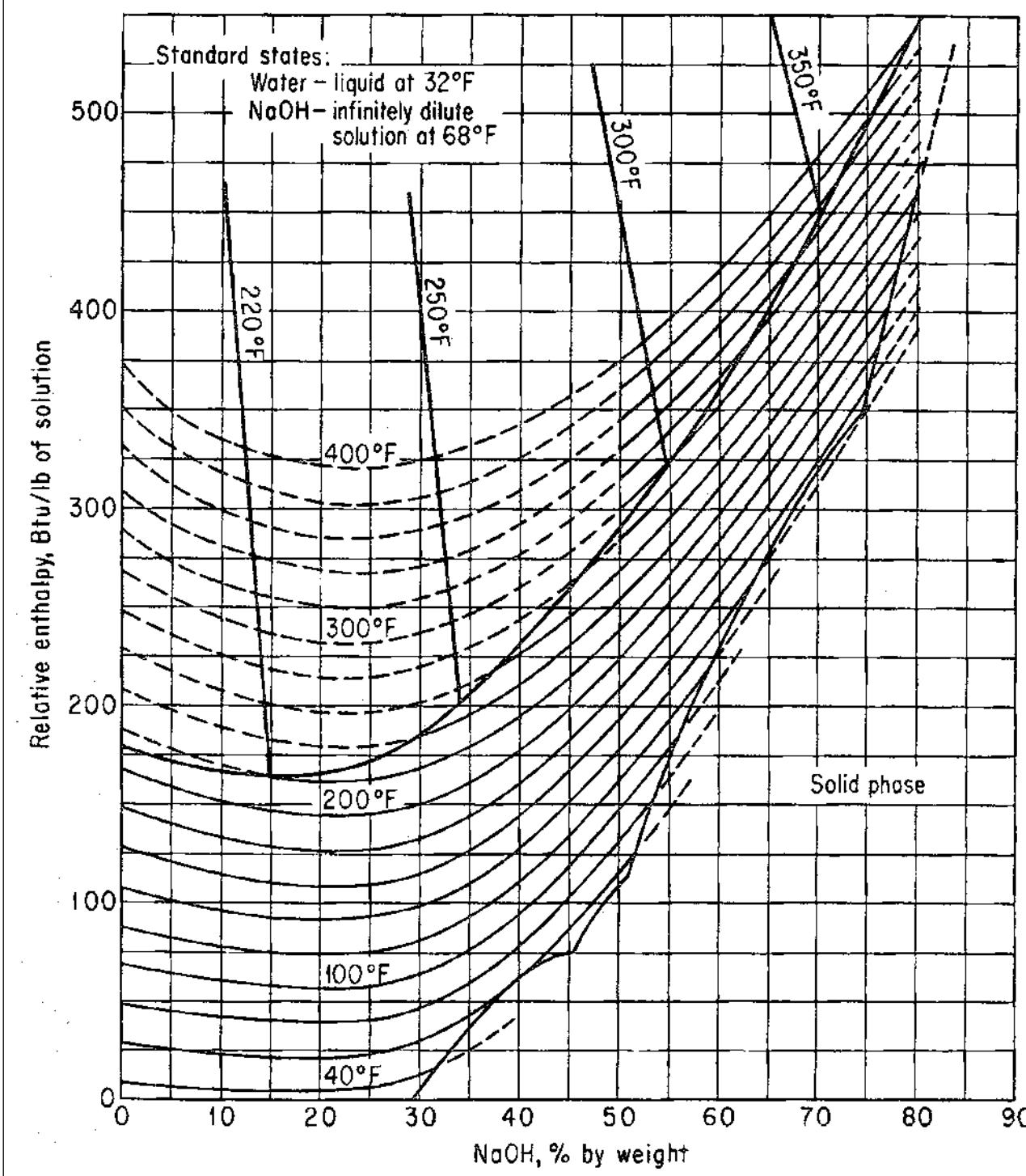
Acqua-Ammoniaca



Acqua-Idrossido di sodio

$$^{\circ}\text{C} = (^{\circ}\text{F} - 32) \cdot 0.556$$

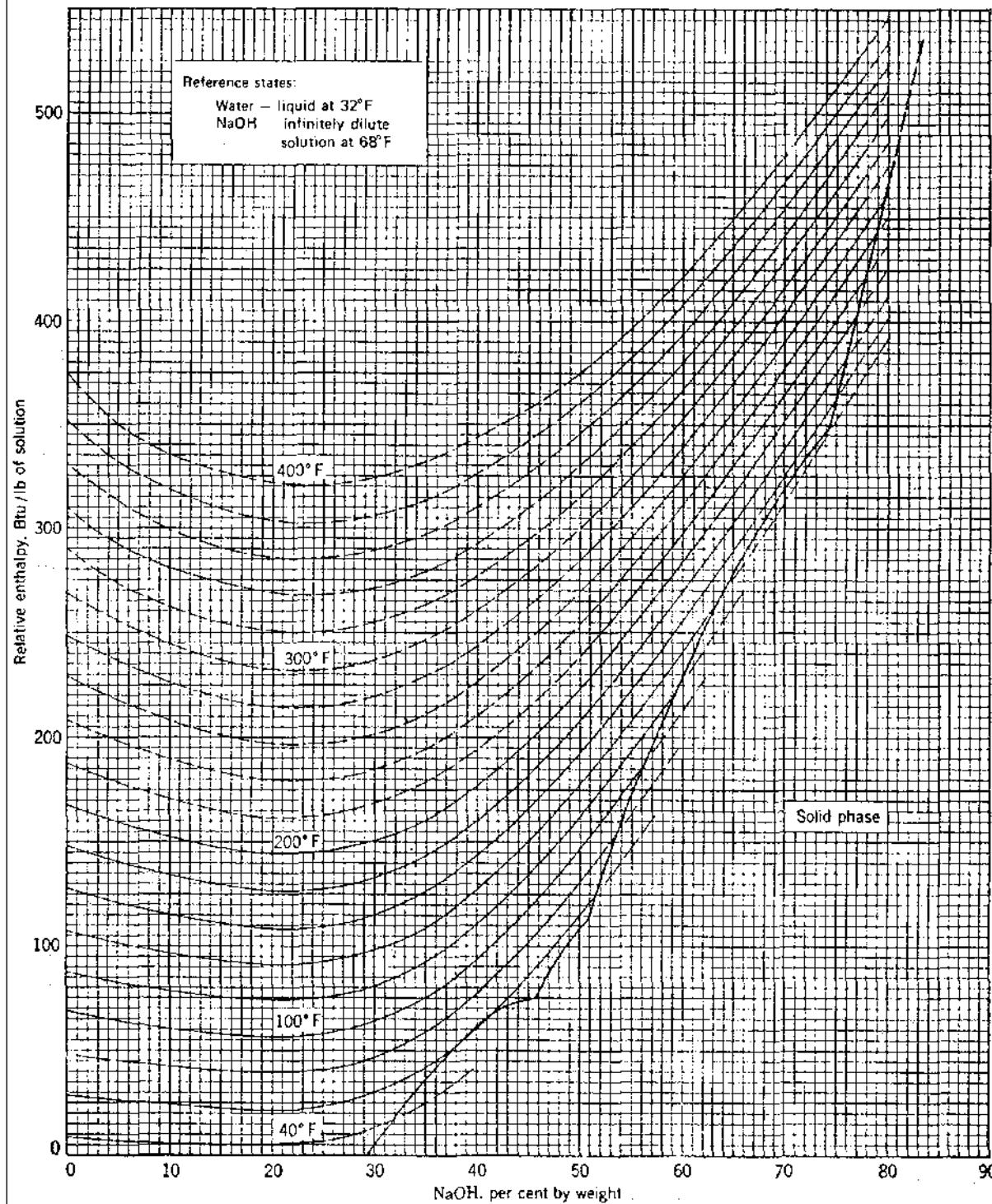
$$1 \text{ Btu/lb} = 2.3 \text{ KJ/Kg}$$



Acqua-Idrossido di sodio 1 atm

$$^{\circ}\text{C} = (^{\circ}\text{F}-32) \cdot 0.556$$

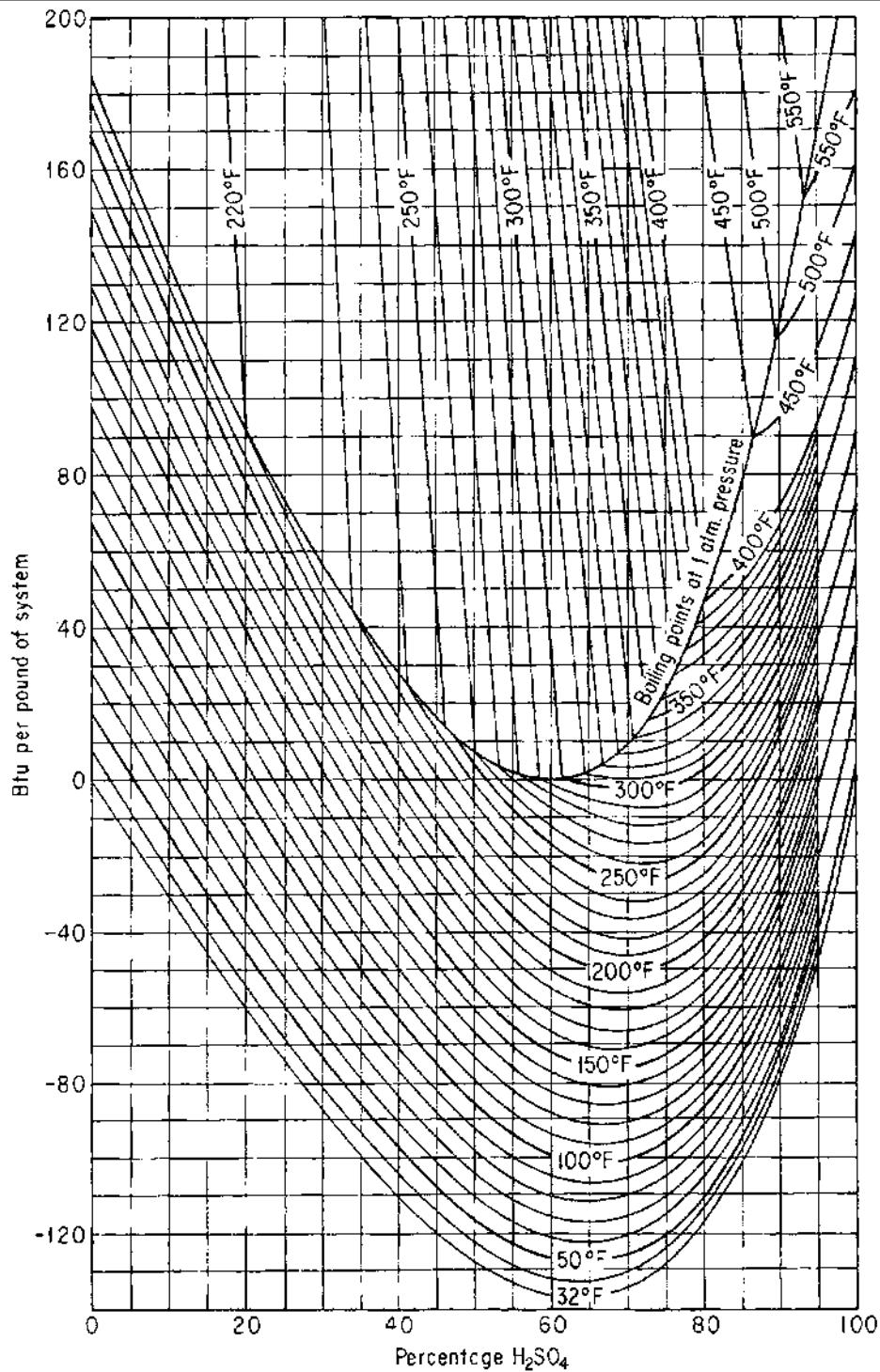
$$1 \text{ Btu/lb} = 2.3 \text{ KJ/Kg}$$



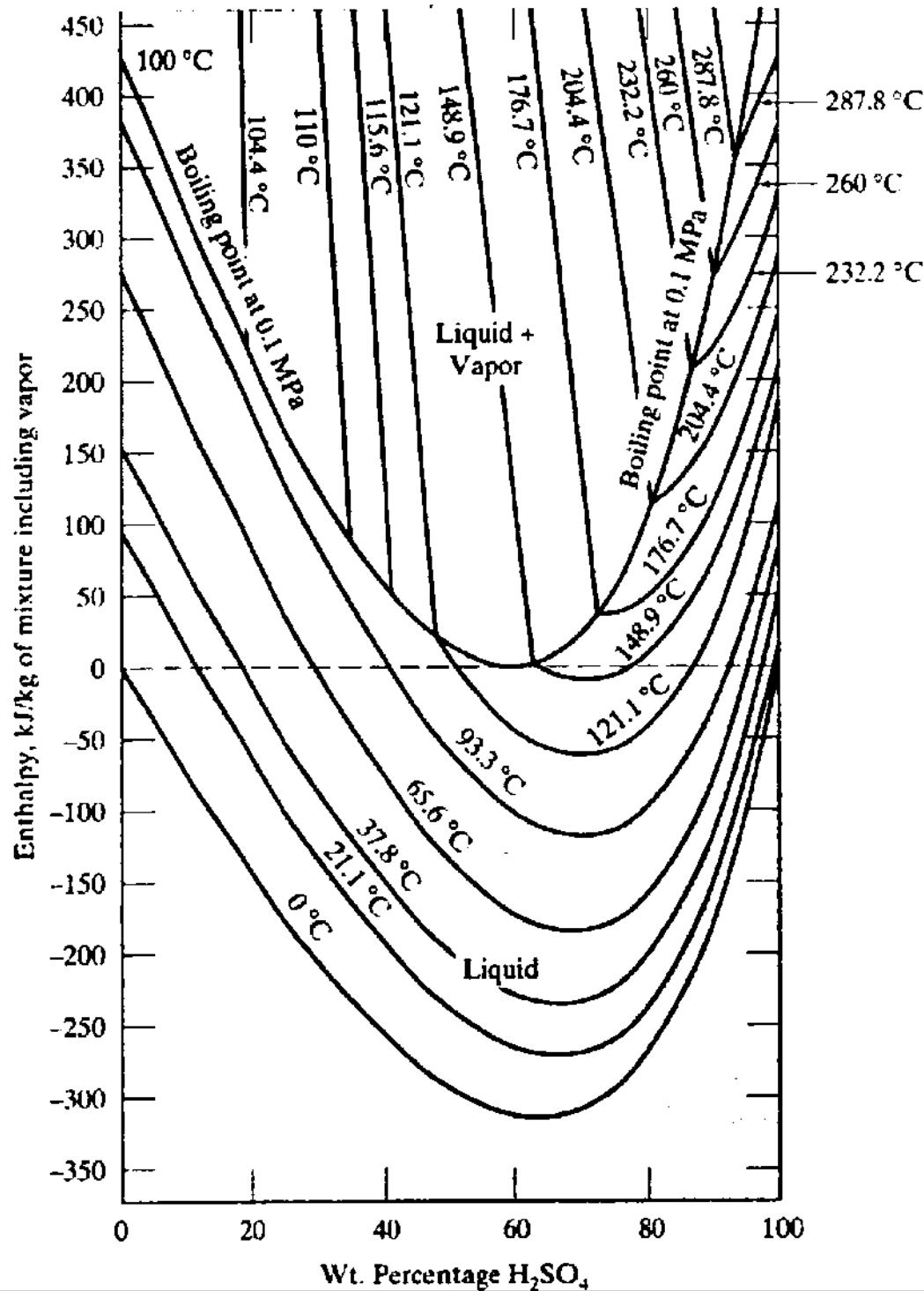
Acqua-Acido solforico 1 atm

$$^{\circ}\text{C} = (^{\circ}\text{F} - 32) / 0.556$$

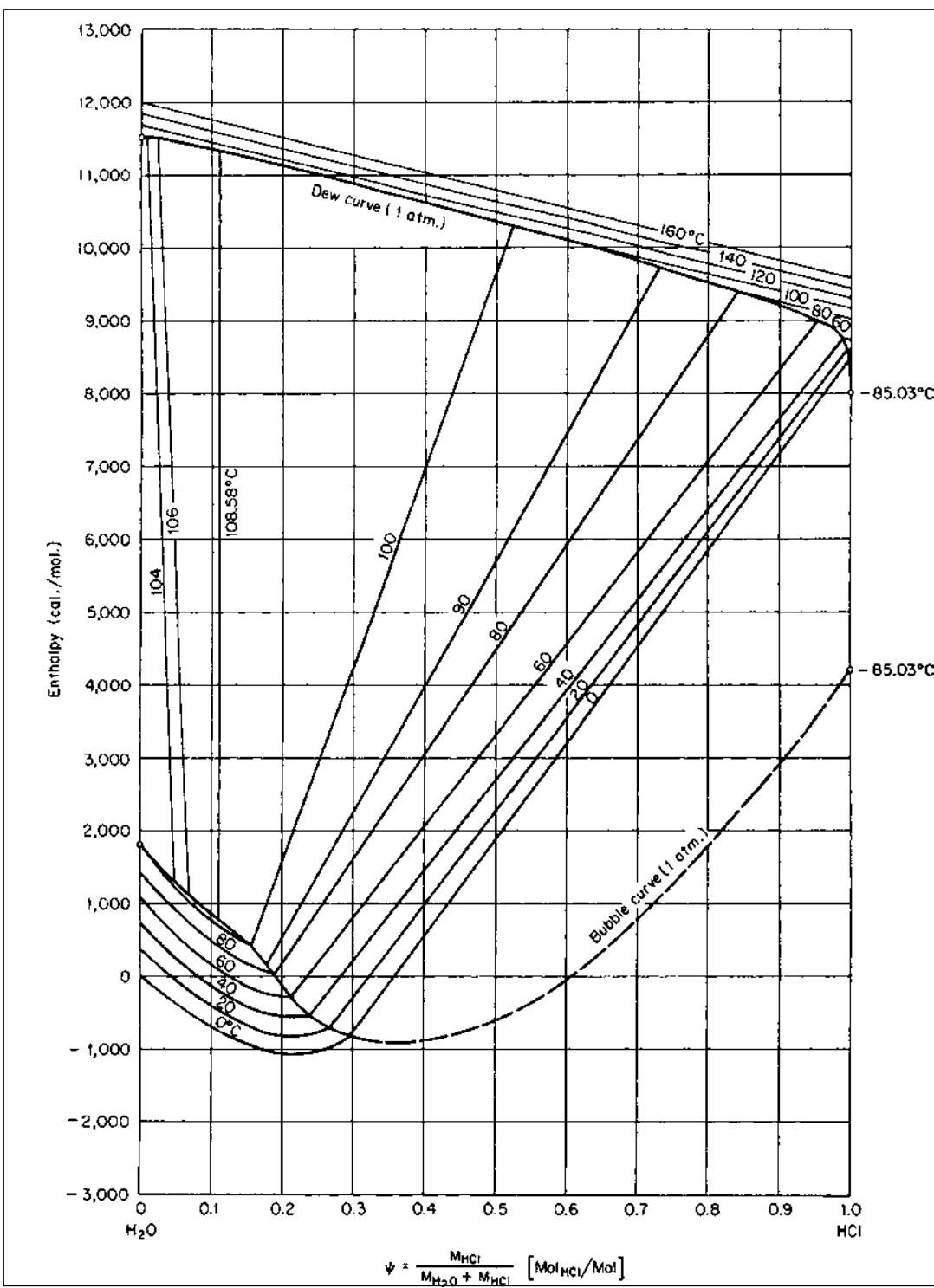
$$1 \text{ Btu/lb} = 2.3 \text{ KJ/Kg}$$



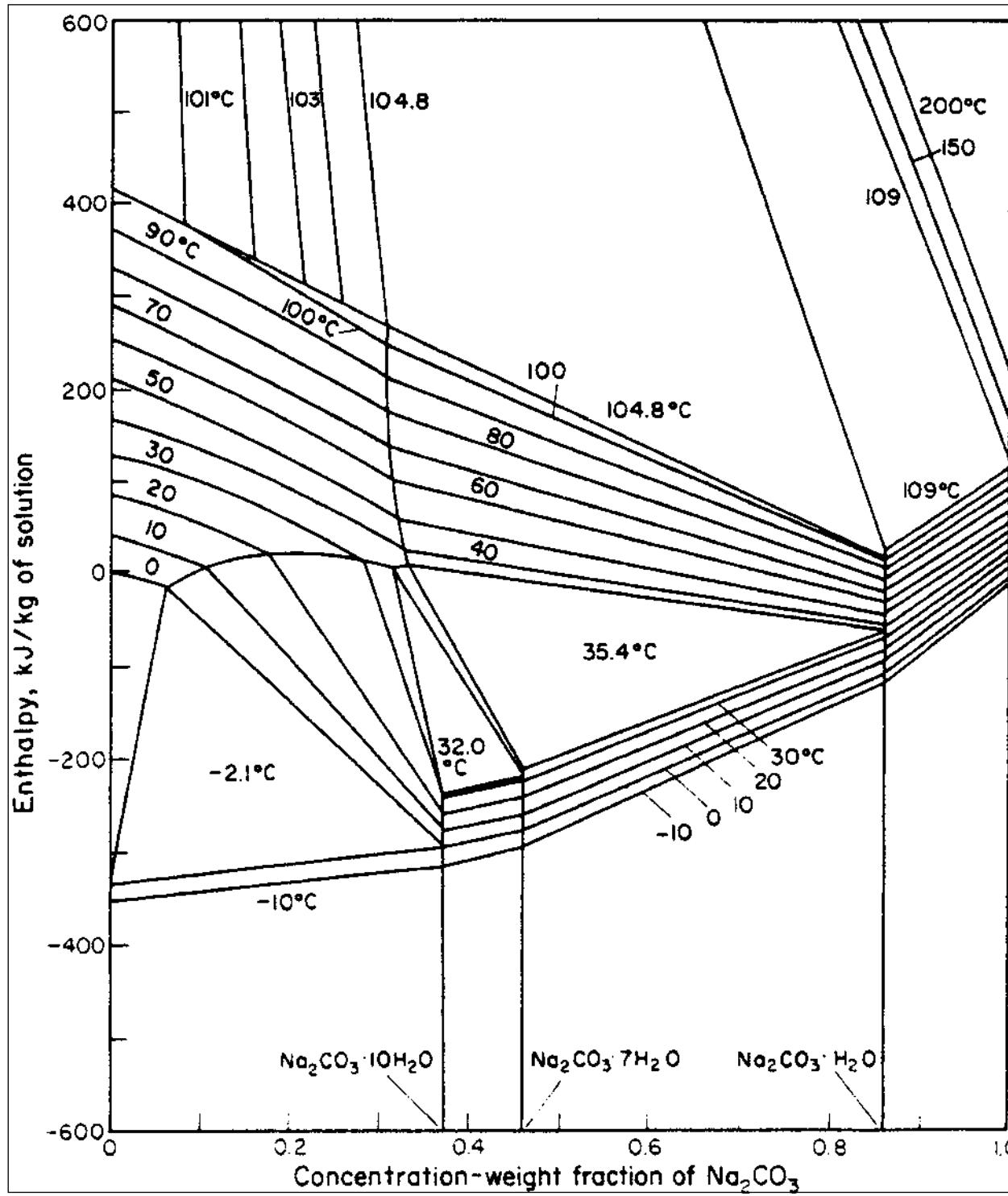
Acqua-Acido solforico 1 atm



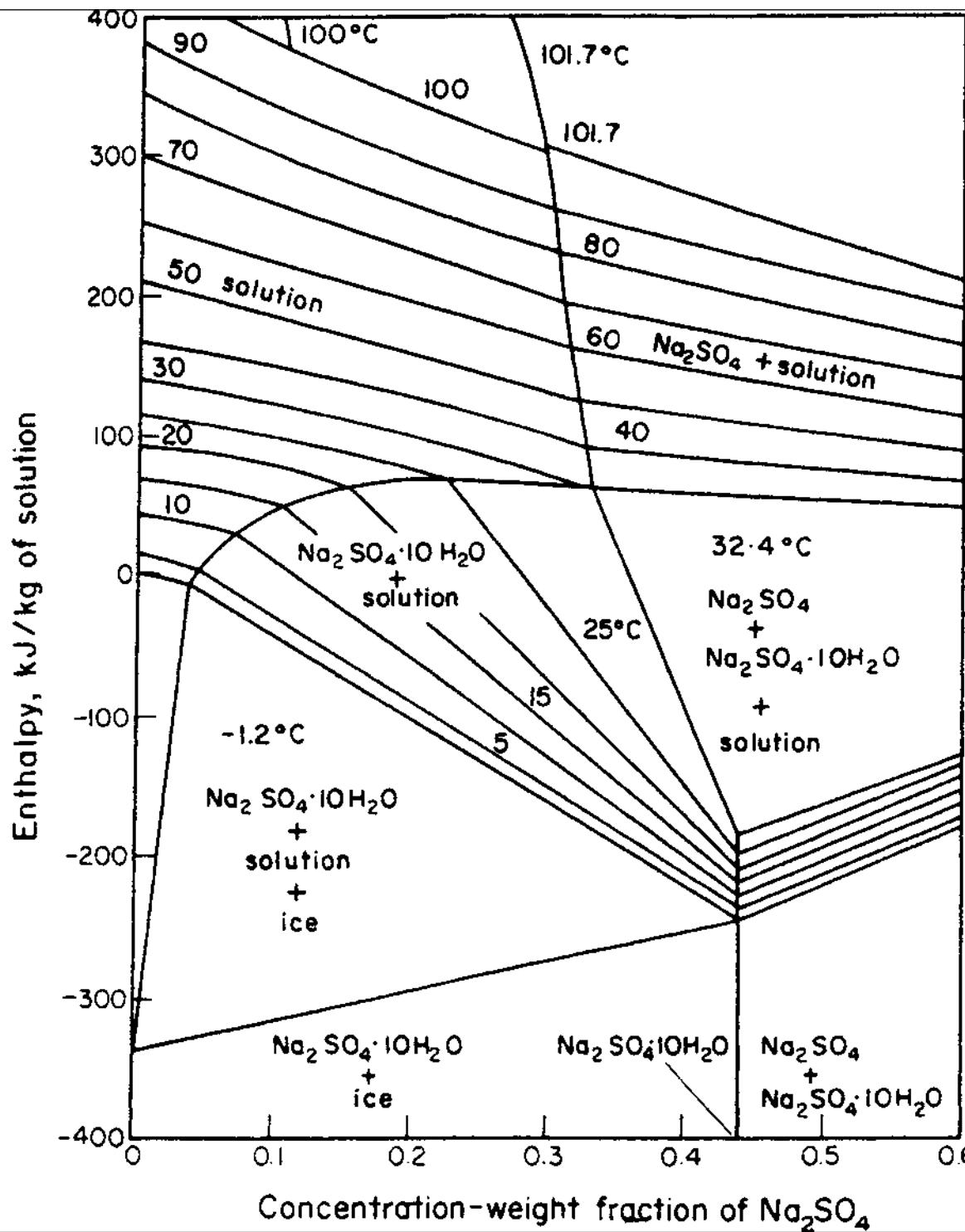
Acqua-Acido cloridrico 1 atm

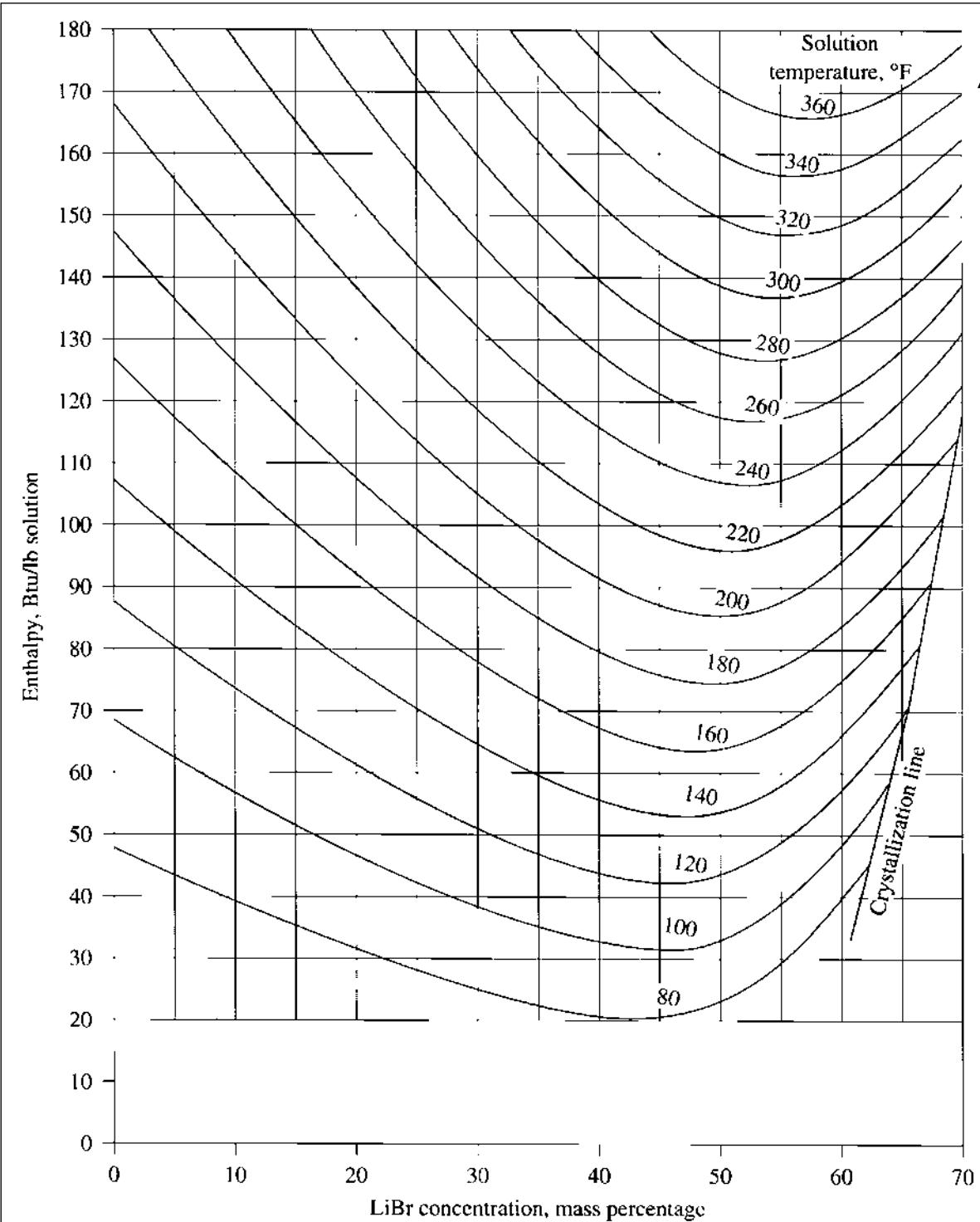


Acqua-Carbonato di sodio 1 atm



Acqua-Solfato di sodio 1 atm





Acqua-Bromuro di Litio 1 atm

$$^{\circ}\text{C} = (^{\circ}\text{F} - 32) \cdot 0.556$$

$$1 \text{ Btu/lb} = 2.3 \text{ KJ/Kg}$$

Acqua-Solfato di Magnesio 1 atm

$$^{\circ}\text{C} = (^{\circ}\text{F}-32) \cdot 0.556$$

$$1 \text{ Btu/lb} = 2.3 \text{ KJ/Kg}$$

